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Produced.  $\text{SO}_3\text{H}$  Benzenesulfonic Acid +  $\text{SO}_3$  The  
Mechanism Of This Reaction Is Similar To That Of The  
Nitration Of Benzene. Concentrated  $\text{H}_2\text{SO}_4$  Is Used In  
An Initial Step To Generate The  $\text{SO}_3\text{H}^+$  Electrophile  
As Shown.  $\text{SO}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{SO}_3\text{H}^+ + \text{HSO}_4^-$  (i)  
Draw A Mechanism For The Reaction Of Benzene With  
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