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3, Benzenesulfonic Acid Is Produced. SO₃ H
Benzenesulfonic Acid + SO₃ The Mechanism Of This Reaction Is Similar To That Of The Nitration Of Benzene. Concentrated H₂ SO₄ Is Used In An Initial Step To Generate The SO₃ H + Electrophile As Shown.
SO₃ + H₂ SO₄ + SO₃ H + HSO₄ - (i) Draw A

Mechanism For The Reaction Of Benzene With $\text{SO}_3 \text{H}^+$
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Add Approximately 10 Drops Of Starch Indicator. A
Blue-black Colour Should Be Seen As The Starch
Reacts With The Residual Iodine. Continue To Add FA 1
1 cm^3 At A Time Until The Blue-black Colour Of The
Starch-iodine Complex Disappears And 3th, 2024

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The Optional Set Texts (2 0 Marks) Answer Any One Of The Following Three Questions. Either (a) The Short

Story Macmillan (E D.), Haifa Day And Other Stories ...

Illustrate Your Answer With References From The Short

Story, 'Half A Day' By Naguib Mahfouz. Or (b) Drama

John Ruganda, Shreds Of Tenderness ... 1th, 2024

Level I Level II Level I Level II Level III Level IV ...

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4 Alkanes And Alkenes Both React With Bromine. (a)
Explain How And Why Bromine Can Be 1th, 2024

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CHEMISTRY 9701/01 1 Hour

10 When Vanadium(II) Compounds Are Dissolved In
Water, The Following Equilibrium Is Established. $V^{2+} + H_2O \rightleftharpoons V^{3+} + H_2O + OH^-$ What Would Alter The
Composition Of The Equilibrium Mixture In Favour Of
The V^{2+} Ions? A Adding An Acid B Adding A Reagent
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