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Chemistry 1A - Unit Three Test Review Chapter 5 Review, P. 183 1. How Is An Ion Different From Its Parent Atom? It Has A Different Number Of Electrons. 2. What Does A Metal Atom Need To Do In Order To Form A Cation? Metal Atoms Need To

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8. Characterize The Solution Formed (strong Acid, Strong, Base, Weak Acid, Weak Base, Buffer Or Neutral) When Equal Volumes Of The Following Are Mixed. Explain.  
 $0.5 \text{ M NaOH} + 0.5 \text{ M HC}_2\text{H}_3\text{O}_2$  The Following Reaction Occurs:  $\text{HC}_2\text{H}_3\text{O}_2 + \text{OH}^- \rightarrow \text{H}_2\text{O} + \text{C}_2\text{H}_3\text{O}_2^-$  - Since The Strong Base Exactly Neutralizes Th Apr 7th, 2024

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INSTRUCTORS Primary Instructor: Prof. Amanda J. Haes (amanda-haes@uiowa.edu; (319) 384 - 3695) Office ... May 1th, 2024

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