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Reaction Follows The Equation: $N_2 + 3H_2 \rightleftharpoons 2NH_3$ (1) $H_0 = 92 \text{ KJ Mol}^{-1}$ $S_0 = 192 \text{ J Mol}^{-1} \text{ K}^{-1}$ 4th, 2024

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