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### **Answer On Question #58182 Chemistry Organic Chemistry ...**

2) Aniline And Acetanilide Stand In A One-to-one Molar Relationship, So 0.0322 Mol



Of Aniline Produces 0.0322 Mol Of Acetanilide. The Amount Of Acetic Anhydride Does Not Play A Role, Since It Is In Excess. Then,  $n(\text{N}) = 0.0322$  .aniline  $n(\text{Aceta II. Ide Mol We Find The Molar Mass Of Acetanilide: } ( \text{N} ) ( ) 12 \ 8 \ 1 \ 9 \ 14 \ 1 \ 16 \ 1 \ 135( ) . 89$  M Aceta ... Jun 1th, 2024

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