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...2. Q. List Three Subatomic Particles Of An Atom. Compare Them On The Basis Of Relative Mass And Charge In A Tabular Form. Ans: Particle Symbol Charge Mass  
Electron  $e^{-}$   $1.60 \times 10^{-19}$  C  $9.1 \times 10^{-31}$  Kg Proton  $+p$  (H<sup>+</sup>)  $1.60 \times 10^{-1}$  May 1th, 2024  
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(c) Three Mole (d) Four Moles (4) Which Of The Following Weighs The Most? (a) One G-atom Of Nitrogen (b) One Mole Of Water (c) One Mole Of Sodium (d) One Molecule Of H<sub>2</sub>SO<sub>4</sub> (5) 5.0 Litre Of Jun 1th, 2024.

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