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$2\text{S}(\text{g})$, $\Delta H^{\circ} = -20.2 \text{ KJ/mol}$ And $\Delta S^{\circ} = +43.1 \text{ J/K}\cdot\text{mol}$. Which Of These Statements Is True? A. The Reaction Is Only Spontaneous At Low Temperatures. B. The Reaction Is Spontaneous At All Temperatures. C. ΔG Mar 29th, 2024

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