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CHAPTER 4 TEST: Atoms, Atomic Theory And Atomic StructureUse The Diagram To The Right To Answer Question 1. _____1. The Atomic Number Of Carbon Is 6, Which Means That Carbon Atoms Always Have 6 A. Ions B. Protons C. Neutrons D. Valence Electrons ____2. In His Investigations Of Air, Henry Cavendish Discovered A Small Bub Feb 1th, 2024Name: Unit 2: Atomic Structure Atomic Structure And ...Then, Write A Sentence Using The Word But Instead Of Writing The Word, Draw A Picture In Its Place. EXAMPLE Word: Noble Gases Definition: The Elements In Group 8A Or 18 Including Helium, Neon, Argon, Krypton, Xenon, And Radon. Sentence: The Are Called Noble Because They Contain 8 Valence Electrons And Do Not React With Other Elements. May 8th, 2024Chapter 4 Atomic Structure Section 4.3 Modern

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