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Stoichiometry: Chapter 5 Thermochemistry: Chapter 6

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Regents Review Stoichiometry A) NO B)NO₂ C) N₂O₃ D) N₂O₅ 25.What Is The Empirical Formula Of A Compound That Contains 30.4% Nitrogen And 69.6% Oxygen By Mass? A) SO B) SO₂ C)SO₃ D) SO₄ 26.A Compound Consists Of 40.% Sulfur And 60.% Oxygen

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Stoichiometry Review Answers 1. A. Na_3PO_4 B. $\text{Ca}(\text{NO}_3)_2$... Use The Following Balanced Equation. Identify The Limiting Reactant When 1.150 Grams Of HgO React With 12.46 Grams Of Cl_2 . Convert To Mol
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Chemistry 11 Stoichiometry Review Worksheet

Name: KEY

Chemistry 11 Stoichiometry Review Worksheet

Name: KEY 33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$

Actual yield = 1.45g Theoretical yield = 5.95g $\text{KClO}_4 \times 1\text{mol}$

$122.55\text{g} \times 2\text{O}_2 / 1\text{KClO}_4 \times 32.00\text{g} / 1\text{mol} = 3.11\text{g O}_2$

% Yield = $1.45\text{g} / 3.11\text{g} \times 100\% = 46.6\%$ 34. $4\text{FeCO}_3 + \text{O}_2$

$\rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$

Percent purity = 62.8%? g pure FeCO_3 Mar 15th, 2024

Chemistry 12, Stoichiometry Review Problems, R. Janssen ...

In A Camping Stove? 2. How Many Moles Of O_2

Molecules Should Be Supplied To Burn 1 Mol Of CH_4 ...

Was At One Time Considered For Use As A Rocket Fuel.

... How Many Formula Units Of Calcium Oxide (CaO) Jan

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