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AnswersSO3 As A Conversion Factor, We Determine

The Mass That This Number Of Moles Of SO3 Has. The

First Step Resembles The Exercises We Did In Section

6.4 "Mole-Mole Relationships In Chemical Reactions".

As Usual, We Start With The Quantity We Were Given:

$3.59 \text{ mol Fe}_2\text{O}_3 \times 3 \text{ mol SO}_3 / 1 \text{ mol Fe}_2\text{O}_3 = 10.77 \text{ mol SO}_3$

The mol Fe₂O₃ units cancel, leaving mol SO₃ ...

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