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CHAPTER 2: Atoms, Molecules And Stoichiometry 2.5 Stoichiometry And Equations Stoichiometry 1) Stoichiometry Is The Proportion Of Things Either Reacting Or Combining. 2) In Compounds, It Refers To The Ratio In Which The Atoms Are Combined Together. For Example, Water,  $\text{H}_2\text{O}$  Has A Stoichiometry Of 2 Hydrogen To 1 Oxygen. 3) It Also Refers To The Reacting Proportions In A Chemical Equation. Feb 2th, 2024

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