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# PREPARATION AND TESTING OF BUFFER SOLUTIONS

67 PREPARATION AND TESTING OF BUFFER SOLUTIONS. P. URPOSE. The Purpose Of The Laboratory Investigation Is To Experimentally Determine (1) PKa (and Thus Ka) Of The Acid In A Buffer And Thus The Buffer Range, (2) Investigate The Buffer Capacity Of Apr 4th, 2024

### Section 19.1. Acid-Base Buffer Solutions

If We Add CH 3COO-SHIFT Net Effect: [H 3O+]
Decreases PH Increases Also, [OH-] Increases POH
Decreases I.e.,a Weak Acid And Its Conjugate Base Or
A Weak Base And Its Conjugate Acid. 19-2 Example:
Addition Of NH 4Cl To NH 3 Jun 2th, 2024

### A Simple And Universal Method For Making Up Buffer Solutions

TBoyer, R F (1986) 'Modern Experimental Biochemistry', Addison- Wesley Publishing Co, Reading, MA 2Bohinski, R C (1983) 'Modern Concepts In Biochemistry', Fourth Edition, Allyn And Bacon, Boston, MA 3Lehninger, A L (1982) 'Principles Of Biochemistry', Worth Publishers, New York Feb 7th, 2024

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