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### **TABLE OF CONJUGATE ACID-BASE PAIRS Acid Base Ka (25 C)**

TABLE OF CONJUGATE ACID-BASE PAIRS Acid Base K A (25 OC) HClO<sub>4</sub> ClO<sub>4</sub><sup>-</sup> - H<sub>2</sub>SO<sub>4</sub> HSO<sub>4</sub><sup>-</sup> - HCl Cl<sup>-</sup> - HNO<sub>3</sub> NO<sub>3</sub><sup>-</sup> - H<sub>3</sub>O<sup>+</sup> + H<sub>2</sub>O H<sub>2</sub>CrO<sub>4</sub> HCrO<sub>4</sub><sup>-</sup> - 1.8 X 10<sup>-1</sup> H<sub>2</sub>C<sub>2</sub>O<sub>4</sub> (oxalic Acid) 1<sup>th</sup>, 2024

### **Chapter 3 Acid-Base Equilibria Acid Base Equilibria ...**

Chapter 3 Acid-Base Equilibria Acid-Base Equilibria Acids And Bases Play A Key Role In A Number Of Environmentally Important Chemical Reactions, Including Weathering, Transport Of Metals In Solution, And CO<sub>2</sub> Atmosphere-water Equilibria. In This Chapter We Will Develop The Concept Of An Acid And A Base, Characterize Strong And Weak Acids, 10<sup>th</sup>, 2024

### **(aq) Acid Base Conjugate Conjugate Acid Base**

Acid Base Conjugate Conjugate . Acid Base . 2) What Is The Strongest Base In The Following Reaction? HNO<sub>3</sub>(aq) + H<sub>2</sub>O(l) ⇌ NO<sub>3</sub><sup>-</sup>(aq) + H<sub>3</sub>O<sup>+</sup>(aq) H<sub>2</sub>O Is The Strongest Base. Strong Acids, Such As HNO<sub>3</sub> Have Weak Conjugate Bases, So NO<sub>3</sub><sup>-</sup> is A Weak Base. H<sub>2</sub>O And 11<sup>th</sup>, 2024

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### **PKa Chart 1 2 Conjugate Acid Conjugate Base Conjugate Acid ...**

Carboxylic Acids Protonated Ketone-7.3 6.37 7 Carbonic Acid Tosic Acid -0.6 Protonated Pyridine 5.2 PKa Chart Conjugate Acid Conjugate Base Conjugate Acid Conjugate Base S T R O N G E S T A C I D S W E A K E S T B A S E S Hydrogen Sulfi 2th, 2024

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In Fact, When A Wine Has \"gone Off\" And Has Acquired A Sour Taste, This Is Due To The Oxidation Of The Ethanol In The Wine To Acetic Acid. (The \"corking\" Of Wine, I.e. Tainting Of The Wine By Compounds Transferred From Or Through The Cork, Is Due To A Totally Different Chemical Process.) Vinegar Derived From Red Or White Wine Is The Most 1th, 2024

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