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Naming Acids 1. NAMING BINARY ACIDS NAMING OXYACIDS1. NAMING BINARY ACIDS: The Name Of The Binary Acid Consists Of Two Words. The First Word Has Three Parts: The "hydro" Prefix The Root Of The Nonmetal Element The "ic" Ending The Second Word Is Always "acid" Examples: HCl = Hydro Chlor Ic Acid = Hydrochloric Acid HBr = Hydro Brom Ic Acid = Hydrobromic Acid 8th, 2024Naming Ionic Compounds - Dorettaagostine.comNaming Ionic Compounds Directions: Name The Following Compounds Using The Stock Naming System. 1. CaCO 3 Calcium Carbonate _____ 2. KCl Potassium Chloride _____ 3. FeSO 4 Iron (II ... Chemistry

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