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1. NAMING BINARY ACIDS: The Name Of The Binary Acid Consists Of Two Words. The First Word Has Three Parts: The "hydro" Prefix The Root Of The Nonmetal Element The "ic" Ending The Second Word Is Always "acid" Examples: HCl = Hydro Chlor Ic Acid = Hydrochloric Acid HBr = Hydro Brom Ic Acid =

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The Simple Ionic Compounds Shown Below: Chemical
Formula Name LiCl Lithium Chloride MgO Valence
Electrons Al 2 O 3 Cyphemical Formula Ba 3 N 2
Barium Nitride MgF 2 Inorganic Compound Na 2 S
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