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1.1 Rate Of Reaction And Dependence On Temperature We Will Once Again Look At The Formation Of Ammonia ( $\text{NH}_3$ ) From Nitrogen And Hydrogen (see Section Chemical Equilibrium Of The Thermodynamics Chapter). This Reaction Follows The Equation:  $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$  (1)  $\Delta H^\circ = -92 \text{ kJ mol}^{-1}$   $\Delta S^\circ = -192 \text{ J mol}^{-1} \text{ K}^{-1}$  Jan 1th, 2024  
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Pantone Violet Pantone Mar 1th, 2024Process 1 Foods Process 2 Foods Process 3  
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Menu/ Flow Analysis Process Worksheet Process 1 Foods Process 2 Foods Process 3  
Foods Foods That Require Foods That Do Not Require A Cook Step (RTE) Foods That  
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