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1 Chemical Reactions 1.1 Rate Of Reaction And Dependence On Temperature We
Will Once Again Look At The Formation Of Ammonia (NH₃) From Nitrogen And
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Reaction Follows The Equation: $N_2 + 3H_2 \rightleftharpoons 2NH_3$ (1) $H_0 = 92 \text{ KJ Mol}^{-1}$ $S_0 = 192 \text{ J Mol}^{-1}$
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Grading: Exam 1 15% Exam 2 15% Exam 3 15% Final Exam 20% Research Paper 10% Homework 20% Class Participation 5% Last Year, The Following Grading Scale Was Used: $A \geq 92$ $A/B \geq 89$ $B \geq 82$ $B/C \geq 80$ $C \geq 71$ $C/D \geq 69$ $D \geq 61$ $F \leq 60$ If Necessary, The Grading Scale May Be Curved Downward. Feb 1th, 2024

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