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 Important About The Variables  $C_p$ ,  $C_v$ ,  $P$ ,  $V$ ,  $T$ ? As A Sample Application, We Can  
 Assume The Ideal Gas Equation Of State.  $V = RT/P \Rightarrow (\partial V / \partial T)_P = R/P$ . Substitution  
 $\Rightarrow \Delta S = C_p \ln(T_2/T_1) - R \ln(P_2/P_1)$  Applying  $\Delta S = 0 \Rightarrow \ln(T_2/T_1) = R/C_p \ln(P_2/P_1)$   
 $\Rightarrow (T_2/T_1) = (P_2/P_1)^{R/C_p}$  23th, 2024 Chemical Biochemical

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