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Example 9-4. Absorption Of  $SO_2$  The Molar Density Of The Feed Gas May Be Determined From The Formula  $(\text{mol/L}) = 12.2P(\text{atm})/T-1(K)$  Where  $P$  And  $T$  Are The Absolute Pressure And Temperature Of The Gas. The Following Data Are Taken For The Feed Gas:  $T=750F$ ,  $P=150$  Psig,  $H=210$  Mm,  $R=82.4$  And For The Outlet Gas,  $R=11.6$ . Determine The (a) Mass Fraction Of  $SO$  Mar 11th, 2024

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