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### **COMPARING QUANTITIES Comparing Quantities 8**

Example 1: A Picnic Is Being Planned In A School For Class VII. Girls Are 60% Of The Total Number Of Students And Are 18 In Number . The Picnic Site Is 55 Km From The School And The Transport Company Is Charging At The Rate Of ` 12 Per Km. The Total Cost Of Refreshments Will Be ` 4280. 2th, 2024

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### **Chemical Reactions And Quantities**

Chemical Reaction • A Chemical Reaction always Involves A Chemical Change Because Atoms Of The Reacting Substances Form New Combinations With New Properties. • For Example, A Chemical Reaction (and Chemical Change) Takes Place When A Piece Of Iron (Fe) Combine 2th, 2024

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## **Chemical Quantities**

1-carbon 2-oxygen Thus One Mole Of  $\text{CO}_2$  Contains Three Times Avogadro's # Of  
Atoms. Example: How Many Atoms Are In 2.12 Mole Of Propane ( $\text{C}_3\text{H}_8$ )? Know: #  
Of Moles = 2.12 Mole  $\text{C}_3\text{H}_8$  1 Mole  $\text{C}_3\text{H}_8$  =  $6.02 \times 10^{23}$  Molecules  $\text{C}_3\text{H}_8$  1  
Molecule  $\text{C}_3\text{H}_8$  = 11 Atoms ( 3 Carbon Atoms & 8 Hydrogen Atoms) 1th, 2024

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## **NOTES: 10.1-10.2 Chemical Quantities (The Mole / Molar ...**

Chemical Quantities -consider Another Word That Means A #: Measuring Doughnuts: A DOZEN! 1 Dozen = 12 Doughnuts (count) 1 Dozen = 500 G Doughnuts (mass) 1 Dozen = 1 Box Doughnuts (volume) Measuring Hydrogen (H<sub>2</sub> Gas): A MOLE! 1 Mole = 6.02 X 10<sup>23</sup> H<sub>2</sub> Molecules (count) 1 Mo 3th, 2024

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" Show All Work And Units ... PRACTICE #3: Calculate The Percent Composition For Each Of The Following 1. When A 13.60-gram 2th, 2024



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