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A: Example Problem: Batch Chemical Reactor

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Ea = Activation Energy [J/mol, Cal/mol] R = Gas Constant = $8.314 \text{ J/mol} \cdot \text{K} = 1.987 \text{ Cal/mol} \cdot \text{K} T = \text{Absolute Temperature [K]}$ The Mass Balance Mass Is A Conservative EntityIV, Hence Given A Control Volume V The Sum Of Mass Flows Entering The System Will Equal The Sum Exiting Minus Jan 13th, 2024

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NOTE: These Practice Problems As Well As NPS Use A "th" Subscript For The Thermal Non-leakage Probability Rather Than Just A "t" As In The DOE Handbook. 3. Calculate The Thermal Utilization Factor, F, For A Homogeneous Reactor. The Macroscopic Absorption Cross Section Of The Fuel Is 0.2028, The Macroscopic Absorption Cross Section Of The Moderator Is 0.0110, And The Macroscopic ... Apr 25th, 2024

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