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Mass Of Empty Crucible Mass Of Crucible & Contents

Before Heating Mass Of Crucible Contents After Heating A. Determine The Formula Of The Hydrate. 20.19.82 G 21.54 G H<sub>2</sub>O Removed - 20.94 G O C. The Formula Of The Hydrate = C. What Is The Name Of The Hydrate? D. What Is Jan 1th, 2024

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PRACTICE PACKET: Unit 7 Moles & Stoichiometry 5

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LESSON 1: Find The Gram Formula Mass Of The

Following: (Show All Work) 1. MgO 5. Ca(OH)<sub>2</sub> 2.

NaHCO<sub>3</sub> 3 6. CH<sub>4</sub> 3. C<sub>6</sub>H<sub>12</sub>O<sub>6</sub> 7. NH<sub>3</sub> 4. Al Apr 3th, 2024

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$\_\_ \text{H}_2 + \_\_ \text{O}_2 \_\_ \text{H}_2\text{O}$  As Written:  $\_\_ \text{Moles}$  Of

Hydrogen Molecules Combine With  $\_\_ \text{Moles}$  Of

Oxygen Molecules To Make  $\_\_ \text{Moles}$  Of Water 2 1 2 2

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GCE In Chemistry (Nuffield) (9086) - Issue 3 - October

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2024

### **Homework 6 Moles Key**

53. Which Particle Has A Diameter Of Approximately 10-15 M? A) Atomic Nucleus B) Atom C) Small Molecule

54. One Of These Statements Is True, Which Is It? A) The Proton Has 1836 Times As Much Mass As The Electron. B) The Proton Has 1839 Times As Much Mass As The E  
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H 1