

Chemistry Chapter 12 Stoichiometry Worksheet

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33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$
Actual yield = 1.45g Theoretical yield = 5.95g
 $\text{KClO}_4 \times 1 \text{ mol} = 122.55 \text{ g} \times \frac{2}{2} = 1 \text{ KClO}_4 \times 32.00 \text{ g} = 3.11 \text{ g O}_2$
% Yield = $\frac{1.45 \text{ g}}{3.11 \text{ g}} \times 100\% = 46.6\%$
34. $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$
Percent purity = 62.8%
pure FeCO_3
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