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Chemistry 11 Stoichiometry Review Worksheet
Name: ___KEY___ 33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$ Actual yield = 1.45g Theoretical yield = 5.95g
 $\text{KClO}_4 \times 1 \text{ mol } 122.55 \text{ g} \times 2 \text{ O}_2 \times 32.00 \text{ g } 1 \text{ mol} = 3.11 \text{ g O}_2$ %Yield = $\frac{1.45 \text{ g}}{3.11 \text{ g}} \times 100\% = 46.6\%$
34. $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$
Percent purity = 62.8%
? g pure FeCO_3 Jan 2th, 2024.
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How Many Moles Of O_2 Molecules Should Be Supplied To Burn 1 Mol Of CH_4 ... Was
At One Time Considered For Use As A Rocket Fuel. ... How Many Formula Units Of
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(stands For Aqueous) → “yield” Symbol For Reactions That Proceed Effectively 100% To Form Product Apr 8th, 2024.

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