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14. Writing The Formulas Of Compounds. A. Binary And Ternary Ionic Compounds : Basically You Have To Work Backward From The Name To The Symbols. For Example Sodium Chloride Involves Sodium And Chlorine And The Formula Is NaCl. What You Need To Be Careful Of Is The Ratio Of Ions In The Compound. The Compound Must Be Neutral (have Equal Positive And Negative Charges) May 20th, 2024

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2. Write The Formulas For The Compounds Formed From These Pairs Of Ions. C.  $\text{Ca}^{2+}$  and  $\text{K}^+$  E.  $\text{O}^{2-}$  and  $\text{Cu}^{2+}$  H.  $\text{Sn}^{2+}$  and  $\text{Cl}^-$  3. Name The Following Binary Ionic Compounds. A.

MnO<sub>2</sub> B. C. CaCl<sub>2</sub> D. SrBr<sub>2</sub> E. F. Chapter May 6th, 2024

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