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(b) He_2 Is Not Stable But He^+_2 Is Expected To Exist. (c) Bond Strength Of N_2 Is Maximum Amongst The Homonuclear Diatomic Molecules Belonging To The Second Period. Thus, Dinitrogen (N_2) Molecule Contains Triple Bond And No Other Molecule Of Second Period Have More Than Double Bond. Hence, Bond Strength Of N_2 Is Maximum Amongst The Homonuclear May 7th, 2024

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Aug 25, 2014 · 5) One Mole Of Nitrogen Gas Contains $(2) \times (6.022 \times 10^{23})$ Nitrogen Atoms. 5) 6) One Mole Of Chlorine Gas Has A Mass Of 35.45 Grams. 6) 7) One Mole Of I_2 Has More Atoms In It Than One Mole Of Na. 7) 8) Six Grams Of Carbon Contains 3.008×10^{23} Atoms. 8) 9) Two Mole(s) Of CO_2 Gas Contains 1 Mole Of Carbon May 17th, 2024

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