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(b)  $\text{He}_2$  Is Not Stable But  $\text{He}^{+}_2$  Is Expected To Exist. (c) Bond Strength Of  $\text{N}_2$  Is Maximum Amongst The Homonuclear Diatomic Molecules Belonging To The Second Period. Thus, Dinitrogen ( $\text{N}_2$ ) Molecule Contains Triple Bond And No Other Molecule Of Second Period Have More Than Double Bond. Hence, Bond Strength Of  $\text{N}_2$  Is Maximum Amongst The Homonuclear Jan 1th, 2024

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Aug 25, 2014 · 5) One Mole Of Nitrogen Gas Contains  $(2) \times (6.022 \times 10^{23})$  Nitrogen Atoms. 5) 6) One Mole Of Chlorine Gas Has A Mass Of 35.45 Grams. 6) 7) One Mole Of  $\text{I}_2$  Has More Atoms In It Than One Mole Of Na. 7) 8) Six Grams Of Carbon Contains  $3.008 \times 10^{23}$  Atoms. 8) 9) Two Mole(s) Of  $\text{CO}_2$  Gas Contains 1 Mole Of Carbon Mar 1th, 2024

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