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SO_3 As A Conversion Factor, We Determine The Mass That This Number Of Moles Of SO_3 Has. The First Step Resembles The Exercises We Did In Section 6.4 "Mole-Mole Relationships In Chemical Reactions". As Usual, We Start With The Quantity We Were Given: $3.59 \text{ Mol Fe}_2O_3 \times 3 \text{ Mol SO}_3 / 1 \text{ Mol Fe}_2O_3 = 10.77 \text{ Mol SO}_3$ The Mol Fe_2O_3 Units Cancel, Leaving Mol SO_3 ... Jun 1th, 2024

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