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Oxide Reacts With  $\text{Br}_2$  And Gives Nitrosyl Bromide As Per Reaction Is Given Below:  
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Dihydrogen At 298 K Are,  $-890.3 \text{ kJ mol}^{-1}$ ,  $-393.5 \text{ kJ mol}^{-1}$ , And  $-285.8 \text{ kJ mol}^{-1}$   
Respectively. Enthalpy Of Formation Of CH Feb 9th, 2024.

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(d) The Feasibility Of ... Mar 14th, 2024 CBSE NCERT Solutions For Class 11  
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And Semicarbazide.  $\text{O} \left[ \text{OH} \right] \text{-----}, 11, / \text{O} \text{)c=ro} \_ \text{-Hrf-N-NH-c -NH Z /'}$ , II  
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