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42 Chapter 2 Environmental Chemistry Section 2.2 Stoichiometry 43 We Already Know That There Are 44 Grams Per Mole Of CO" So We Do Not Need To Recalculate That. Two Moles Of Butane ($2 \text{ Mol} \times 58 \text{ G/mol} = 116 \text{ G}$) Yields 8 Moles Of Carbon Dioxide ($8 \text{ Mol} \times 44 \text{ G/mol} = 352 \text{ G CO}_2$) , So We Can Set Up The Following Proportion: 4th, 2024

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