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LEWIS STRUCTURES General Rules For Drawing Lewis Structures

ASSIGNING FORMAL CHARGE ON ATOMS IN COMPOUNDS Formal Charge = Number Of Valence Electrons In The Free Atom (the Group Number) — Electrons Surrounding The Atom In The Compound (count Each Electron In The Lone Pair Electrons And One Electron For Each Bond). 1th, 2024

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(filled Valence Shell Of 8 Electrons) • Electron-Dot Diagrams Of The Atoms Are Combined To Show The Covalent Bonds • Covalently Bonded Atoms Form MOLECULES Methane CH₄ • This Is The Finished Lewis Dot Structure April 10th, 2019 - Chapter 8 Covalent Bonding And Molecular Structure 7th, 2024

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