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 Nitrogen And Hydrogen (see Section Chemical
 Equilibrium Of The Thermodynamics Chapter). This
 Reaction Follows The Equation: $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ (1)
 $\Delta H^\circ = -92 \text{ kJ/mol}$ $\Delta S^\circ = -192 \text{ J/mol K}$ Feb 2th, 2024.
 Chemical Engineering, CM3510: Chemical Reaction
 ...Grading: Exam 1 15% Exam 2 15% Exam 3 15% Final
 Exam 20% Research Paper 10% Homework 20% Class
 Participation 5% Last Year, The Following Grading
 Scale Was Used: $A \geq 92$ $A/B \geq 89$ $B \geq 82$ $B/C \geq 80$ $C \geq 71$
 $C/D \geq 69$ $D \geq 61$ $F \leq 60$ If Necessary, The Grading
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