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Objective: To Determine The Specific Heat Of A Metal.  
Theory: The Amount Of Heat Energy That Is Required To Raise The Temperature Of One Gram Of A Substance By One Degree Celsius Is Called The Specific Heat Capacity, Or Simply The Specific Heat, Of That Substance. Water, For Instance, Has A Specific Jan 23th, 2024.

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Lab Report: Calorimetry And Hess's Law  
Calorimetry And Hess's Law Page 1 Of 4  
Metal + HCl Reaction Name: Date: Lab Partner: Lab Section: Lab Report: Calorimetry And Hess's Law  
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Mass Of Dry, Empty Calorimeter Mass Of Calorimeter May 29th, 2024  
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Day 3 Calorimetry Lab - Travellin - HomeLab 6b, Page 4 HASPI Medical Chemistry Unit 6: Energy Using Hess' Law You Can Reverse The Second Reaction And Add It To The First In Order To Calculate The  $\Delta H$  Of The Goal Reaction. First We Will Calculate The  $\Delta H$  For Reaction 1 And Reaction 2 Using Calorimetry Jan 15th, 2024

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