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Electrolyte Cases, Treated By An Ionization Constant,
Differ From The Strong Electrolyte Cases Whose
Essentially Complete Dissociation Makes It Possible To
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Salt Solutions May Be Acidic, Basic, Or Neutral,
Depending On The Original Acid And Base That Formed
The Salt. Strong Acid + Strong Base Neutral Salt
Strong Acid + Weak Base Acidic Salt ... 1. Complete
The Following Hydrolysis Reactions For These Ions.
 $F^{-}(aq) + H_2O$ 8th, 2024

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