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Answer On Question #60099, Chemistry / Inorganic Chemistry  
1. 1.00g Of Carbon Is Combusted In A Limited Supply Of Pure Oxygen. 0.50g Of The Carbon Combusts To Form CO<sub>2</sub> And 0.50 G Of The Carbon Combusts To Form CO. The Resultant Mixture Of CO<sub>2</sub> And CO Is Passed Through Excess Na Feb 1th, 2024

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