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(2) To Determine The Proper Number Of Sig. Figs When Multiplying Or Dividing The Measurement With The Least Number Of Sig. Figs Is Boss G. If All This Is Confusing, Use The (A)tlantic – (P)acific Rule. If The Decimal Point Is (A)bsent – Start Counting Significant Figures From The Atlantic 5th, 2024

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Element J Ha	as 27 Protons An	id 32 Neutro	ns. Ele	ment L Has 27	Protons And 33
Neutrons	_ 3. Element X Ha	as 17 Protor	s And	18 Neutrons. E	lement Y Has 18
Protons And 17	Neutrons	4. Element	Q Has	56 Protons And	d 81 Neutrons.
Element F Has	56 Protons And	82 Neutrons	. <u></u>	5. Element T F	las An Atomic
Number 20 And	d Mass Number 4	40. Flement	7 61	th. 2024	

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D. Uranium-235: 92 Protons, 143 Neutrons E. Carbon-12: 6 Protons, 6 Neutrons 3. Most Of An Atom's Mass Is Concentrated In The Nucleus. The Number Of Electrons And Protons Is The Same But Electrons Are So Light They Contribute Very Little Mass. The Mass Of The Proton Is 1,835 Times The Mass Of The Electron. Neutrons Have A Bit More Mass Than ... 18th, 2024

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