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2 → CaSiO_3 B. $3\text{Fe}_2\text{O}_3 + \text{CO} \rightarrow 2\text{Fe}_3\text{O}_4 + \text{CO}_2$ C. $\text{FeO} + \text{SiO}_2 \rightarrow \text{FeSiO}_3$ D. FeO

→ Fe + 1 2 O 2 A. A B. D C. C And D D. A And D Answer: C Solution: In Metallurgy Of Iron, CaO Is Used As Flux Which Is Used To Remove The Impurities Of SiO 2 , CaO + SiO 2 → CaSiO 3 . Also Here Fe 2 O 3 Is Reduced By CO To Fe 3 O 4 Which Is Further Reduced To ... 1th, 2024

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