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 $2 \rightarrow$ CaSiO 3 B. 3Fe 2 O 3 + CO → 2Fe 3 O 4 + CO 2 C. FeO +SiO 2 → FeSiO 3 D. FeO → Fe + 1 2 O 2 A. A B. D C. C And D D. A And D Answer: C Solution: In Metallurgy Of Iron, CaO Is Used As Flux Which Is Used To Remove The Impurities Of SiO 2 , CaO + SiO 2 → CaSiO 3 . Also Here Fe 2 O 3 Is Reduced By CO To Fe 3 O 4 Which Is Further Reduced To ... 2th, 2024

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