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Ionic Bonds Properties Of Ionic Compounds (salts)Naming Binary Ionic Compounds Contain 2 Different Elements Name The Metal First, Then The Nonmetal With The Ending -ide. Use Name Of A Metal With A Fixed Charge These Are: Groups 1A, 2A, 3A And Ag +, Zn 2+, And Cd 2+ Examples: NaCl Sodium Chloride Znl 2 Zinc Iodide Al 2O3 Aluminum Oxide Lear May 2th, 20247.2 Ionic Bonds And Ionic Compounds -Henry County SchoolsThe Ionic Compound Magnesium Chloride Contains Magnesium Cat-ions (Mg2) And Chloride Anions (Cl). In Magnesium Chloride, The Ratio Of Magnesium Cations To Chloride Anions Is 1:2 (one Mg2 To Two Cl). So Its Formula Unit Is MgCl 2. Because There Are Twice As Many Chloride Anions (each With A 1 Mar 1th, 2024Ionic Bonding & Ionic Compounds Multiple Choice Review ...A) Metals, Metals B) Nonmetals, Nonmetals C) Metals, Nonmetals D) Nonmetals, Metals E) Metalloids, Metalloids 4) Which Species Below Is The Nitride Ion? A) Na+ B) NO 3-C) NO 2-D) NH 4 + E) N3-5) When A Metal And A Nonmetal React, The Tends To Lose Electro Apr 2th, 2024.

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