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Show Your Work, Including Proper Units, To Earn Full Credit. 1. Silver And Nitric Acid

React According To The Following Balanced Equation: $3 \text{Ag}(s) + 4 \text{HNO}_3(aq) \rightarrow 3$

$\text{AgNO}_3(aq) + 2 \text{H}_2\text{O}(l) + \text{NO} \dots$ 1th, 2024

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Chemistry Supplemental Stoichiometry Problems Answer Key Stoichiometry Is A

Chemical Section Involving The Use Of Relationships Between Reagents And / Or

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Greek, Stoikhein Means Element And Metricity Means Measurement, Then Steichiometry Literally Translated Means ... 1th, 2024

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Chemistry 11 Stoichiometry Review Worksheet Name: KEY

Chemistry 11 Stoichiometry Review Worksheet Name: ___KEY___ 33. $\text{KClO}_4 \rightarrow 2\text{O}_2$

$2\text{KClO}_4 \rightarrow 2\text{KCl} + 3\text{O}_2$ Actual yield = 1.45g Theoretical yield = 5.95g $\text{KClO}_4 \times 1\text{mol} \times 122.55\text{g/mol} \times 2 = 245.10\text{g}$ $\text{KClO}_4 \times 32.00\text{g/mol} = 3.11\text{g O}_2$ %Yield = $\frac{1.45\text{g}}{3.11\text{g}} \times 100\% = 46.6\%$

$4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$ Percent purity = 62.8%? g pure FeCO_3 1th, 2024

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Basic Stoichiometry Phet Lab Answer Key Page 15/28 Read Online Basic Stoichiometry Phet Lab Answer Key (ANSWER 386.3g Of LiNO_3) 4) Using The Following Equation: $\text{Fe}_2\text{O}_3 + 3\text{H}_2 \rightarrow 2\text{Fe} + 3\text{H}_2\text{O}$. Calculate How Many Grams Of Iron Can Be Made From 16.5 Grams Of Fe_2O_3 By The Following Equation. 1th, 2024

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