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$2\text{KClO}_4 \rightarrow 2\text{KCl} + 2\text{O}_2$
Actual yield = 1.45g Theoretical yield = 5.95g
 $\text{KClO}_4 \times 1\text{mol} = 122.55\text{g}$
 $2 \times 122.55\text{g} = 245.1\text{g}$
 $\frac{1.45\text{g}}{245.1\text{g}} \times 100\% = 0.59\%$
 $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$
Percent purity = 62.8%

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