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Arrange The Following Polymers In Increasing Order Of Their Intermolecular Forces (i) Nylon 6, 6, Buna-S, Polythene (ii) Nylon 6, Neoprene, Polyvinyl Chloride. Solution: On The Basis Of Intermolecular Forces Of Attraction, The Polymers Can Be Classified Into Elastomers, F Feb 15th, 2024.

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Mendeleev Arranged The Elements In His Periodic Table According To Their Atomic Weight Or Mass. He Arranged The Elements In Periods And Groups In Order Of Their Increasing Atomic Weight. He Also Kept The Elements With Similar Properties In The Same Group. But, He Found Out That If The Elements Were Feb 17th, 2024  
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 $\Delta G = -nFE$   $E = \frac{RT}{nF} \ln \frac{[\text{H}^+]}{[\text{H}_2]}$  Class- XII-CBSE-Chemistry Electrochemistry Practice More On Electrochemistry Page - 3 [www.embibe.com](http://www.embibe.com) ... Has  $E_{\text{cell}} = 0.236 \text{ V}$  At 298 K. Calculate The Standard Gibbs Energy And The Equilibrium Constant O Mar 4th, 2024  
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