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Tert-butoxide Ion Is Very Bulky, Or Sterically Hindered,  
And Is Less Capable Of Interacting With The Solvent.  
The Ethoxide Ion Is Not As Sterically Hindered So It Can  
Accommodate More Solvent Interactions. As A Result  
Ethoxide Is Better Solvated And Is Therefore More  
Stable Than Tert-butoxide (Figure 3.7). May 12th,  
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Organic ...Cules Of Acetic Acid And 2 Molecules Of  
Water To Form Indigo. Molecules Of Acetic Acid And 2  
Molecules Of Water To Form In-digo. Fig. 3: Overall  
Reaction Equation Of Indigo Synthesis. The First Step  
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Dot Structure)! Both Atoms Now Have 8 Electron Jan  
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And An Alcohol. • What Is The Maximum Number Of  
H-bonds An ... • Without The Crown Ether, The  
Solubility Of KF In Benzene ... • Consider A  
Retrosynthetic Disconnect On The T-butyl Side. • It Is  
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Exam Organic Chem Name (last ...C)

1,1-dibromo-2-chloro-2-fluoropropane

D)1,1-dibromo-2,2 Dichlorofluoropropane 12 Ethyne Is  
A(n) A) Alkane B) Alkene C) Alcohol D) Aromatic E)

Alkyne 13 How Many Alkyne Isomers Are Possible With  
The Molecular Formula  $C_6H_{10}$ ? A) 3 Or Less B) 4 C) 5  
D) 6 E) 7 Or More 14 Alkanes Have The General  
Formula A)  $C_nH_{2n+2}$  B)  $C_nH_{2n}$  C)  $C_nH_{2n-2}$  D)  $C_nH_n$  E  
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