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Sulphite Chloride Sulphite Chloride (b) This Reaction Is Also Known As Double Displacement R Eaction (c)  $\text{BaSO}_3$  Is A Salt Of A Weak Acid ( $\text{H}_2\text{SO}_3$ ), Therefore Dilute Acid Such As HCl Decomposes Barium Sulphite To Pr Oduce Sulphur Dioxide Gas Which Has The Smell Of Bur Ning Sulphur.  $\text{BaSO}_3 (\text{s}) + 2\text{HCl} (\text{aq}) \rightarrow \text{BaCl}_2 + \text{H}_2\text{O} + \text{SO}_2 (\text{g})$  White Ppt. 3th, 2024

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