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### **Study Of Reaction Rates: Clock Reaction Lab**

Chemical Reaction- Called A Clock Reaction- Will Be  
Used To Determine Quantitatively The Influence Of  
Concentration On Rate. The First Reaction Used Is The  
Oxidation Of Iodide Ions By Hydrogen Peroxide In  
Aqueous Solutions:  $2\text{I}^-(\text{aq}) + \text{H}_2\text{O}_2(\text{aq}) + 2\text{H}^+(\text{aq}) \rightarrow \text{I}_2(\text{aq}) + 2\text{H}_2\text{O}(\text{l})$ , 2024

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