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2 Discussion Points And Possible Answers Move To Pages 1.2 - 1.4. Have Students Answer Questions 1-3 On The Handheld, The Activity Sheet, Or Both.
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1) $1 \text{C}_3\text{H}_8 + 5 \text{O}_2 \rightarrow 3 \text{CO}_2 + 4 \text{H}_2\text{O}$ 2) $2 \text{Al} + 1 \text{Fe}_3\text{N}_2 \rightarrow 2 \text{AlN} + 3 \text{Fe}$ 3) $2 \text{Na} + 1 \text{Cl}_2 \rightarrow 2 \text{NaCl}$ 4) $2 \text{H}_2\text{O}_2 \rightarrow 2 \text{H}_2\text{O} + 1 \text{O}_2$ 5) $1 \text{C}_6\text{H}_{12}\text{O}_6 + 6 \text{O}_2 \rightarrow 6 \text{H}_2\text{O} + 6 \text{CO}_2$ 6) $4 \text{H}_2\text{O} + 7 \text{CO}_2 \rightarrow 1 \text{C}_7\text{H}_8 + 9 \text{O}_2$ 7) $2 \text{NaClO}_3 \rightarrow 2 \text{NaCl} + 3 \text{O}_2$ 8) $4 (\text{NH}_4)_3\text{PO}_4 + 3 \text{Pb}(\text{NO}_3)_4 \rightarrow 1 \text{Pb}_3(\text{PO}_4)_4 + 12 \text{NH}_4\text{NO}_3$ Jan 11th, 2024.

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5 Q5. Can You Use Fractions As Coefficients In The Chemical Equations? Answer: Since There Cannot Be Half Of A Molecule Or Half Of An Atom, The Final Set Of Coefficients Has
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C. Nuclear Reactions Often Produce Large Amounts Of Energy Because Small Amounts Of Mass Are Converted Into Energy (see Einstein's Famous Equation, $E=mc^2$)
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30. Sr Predict The Decay Pathway For 90. (Strontium-88 Is The Most Abundant Stable Isotope For ...
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The Balancing Force Is 180o Anticlockwise Of This So The Balancing Mass Must Be Placed At An Angle Of 208o.
D.J.Dunn 12 Here Is The Solution To Worked Example No.2 Using The Same Method. Mass Radius M R X Mrx θ A MA 60 60 MA 0 0 Unknown B 5 75 375

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