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Chapter 14 Chemical Kinetics 3 Chemical Kinetics Factors That Affect Reaction Rates • Physical State Of The Reactants In Order To React, Molecules Must Come In Contact With Each Other. If The Reaction Is Happening Between A Solid And A Liquid It Will React Only On The Surface. The More Homogeneous The Mixture Of Reactants, The Faster The Molecules Can React. Jan 2th, 2024 Chemical Kinetics Reaction Rates 5 The Overall Order Of A Reaction Is The Sum Of The Individual Orders: $Rate (M s^{-1}) = k[A][B]^{1/2}[C]^2$ Overall Order: $1 + \frac{1}{2} + 2 = 3.5 = 7/2$ Or Seven-halves Order Note: When The Order Of A Reaction Is 1 (first Order) No Exponent Is Written. Units For The Rate Constant: The Units Of A Rate Constant Will Change Depending Upon The Overall May 10th, 2024 Chemical Kinetics And Photochemical Data For Use In ... iii ABSTRACT This Is The Fourteenth In A Series Of Evaluated Sets Of Rate Constants And Photochemical Cross Sections Compiled By The NASA Panel For Data Evaluation. Feb 10th, 2024.

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