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Four Of The Remaining 2p X And 2p Y Orbitals Must Make The  $\pi$  And  $\pi^*$  Bonding And

Antibonding Orbitals. The "left Over" 2p X Orbital Goes Into An Approximately Nonbonding Orbital. (ie The Three 2p X Orbitals Form The Combinations +++, +.-, And +-+). (2 Marks For 2p X As B 1 And 2p Y As B 2, Both Of Which Should Be Read Straight From The Character Table) (d) How Many Valence Electrons ... Jun 16th, 2024

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