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Dependence Of The Solubility On Temperature ΔH_{diss} Is The Partial Molar Enthalpy And ΔS_{diss} Is The Partial Molar Entropy Of The
I-th Component. Substituting This Into Eq. 3, We Get The Correlation $\ln x_{\text{sat}} = \frac{\Delta H_{\text{diss}}}{RT} + \Delta S_{\text{diss}}$
M,solid R (6) This Includes Partial Molar Quantities, Values That May Also Be Computed May 1th, 2024 Solubility Temperature
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