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1.1 Rate Of Reaction And Dependence On Temperature We Will Once Again Look At  
The Formation Of Ammonia ( $\text{NH}_3$ ) From Nitrogen And Hydrogen (see Section  
Chemical Equilibrium Of The Thermodynamics Chapter). This Reaction Follows The  
Equation:  $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$  (1)  $H_0 = 92 \text{ KJ Mol}$   $S_0 = 192 \text{ J Mol K}$  May 1th,  
2024Chemical Engineering, CM3510: Chemical Reaction ...Grading: Exam 1 15%  
Exam 2 15% Exam 3 15% Final Exam 20% Research Paper 10% Homework 20%  
Class Participation 5% Last Year, The Following Grading Scale Was Used:  $A \geq 92$   $A/B$   
 $\geq 89$   $B \geq 82$   $B/C \geq 80$   $C \geq 71$   $C/D \geq 69$   $D \geq 61$   $F \leq 60$  If Necessary, The Grading  
Scale May Be Curved Downward. Jun 1th, 2024Octave Levenspiel Chemical Reaction  
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