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ENTHALPY AND ENTROPY OF A BORAX SOLUTION

Since Borate Ion Is A Base, Its Concentration Can Easily Be Determined By A Simple Acid-base Titration. The Endpoint Is Signaled By The Color Change Of Bromocresol Green Indicator. (7) $B_4O_5(OH)_4^{2-} + 2H^+ + 3H_2O \rightarrow 4B(OH)_3$ Example: A Student Prepared 5.00-mL Of Borax Solution At 59.2°C. He Diluted The Solution In A May 1th, 2024

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Higher-Order Poincare´ Sphere, Stokes Parameters, And The Angular Momentum Of Light Giovanni Milione,1 H.I. Sztul,1 D.A. Nolan,2 And R.R. Alfano1,* 1The City College Of New York Of The City University Of New York, 160 Convent Avenue New York, New York 10031, USA 2Corning Incorporated, Sullivan Park, Corning, New York 14831, USA (Received 21 March 2011; Published 25 July 2011) Jun 1th, 2024

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That Black Holes Thermally Radiate And Calculated The Black-hole Temperature. The Main Feature Of The Bekenstein–Hawking Entropy Is Its Proportionality To The Area Of The Black-hole Horizon. This Property Makes It Rather Different From The Usual Entropy, For Example The Entropy Of A Thermal Gas In A Box, Which Is Proportional To The Volume. In 1986 Bombelli, Koul, Lee And Sorkin [23 ... Jan 1th, 2024

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